

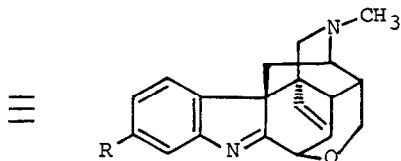
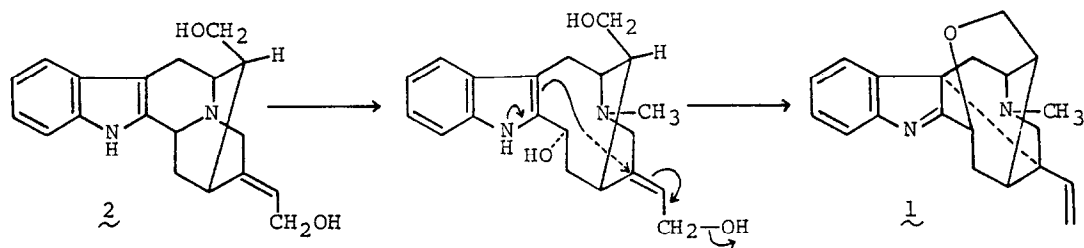
BIOMIMETIC SYNTHESIS OF KOUMINE SKELETON: PARTIAL SYNTHESIS OF  
 11-METHOXYKOUMINE ( GELSEMIUM-TYPE ALKALOID ) FROM 18-HYDROXYGARDNERINE

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Summary: 11-Methoxykoumine (13) was derived from 18-hydroxygardnerine (4) through C-C intramolecular bond formation by Pd<sup>0</sup> catalysis between the indole part and allylic cation in an indole alkaloid. This is first example of the proof for the biogenetic route of koumine skeleton.

Koumine (1) is one of Gelsemium alkaloids ( Loganiaceae ), and its complicated indole alkaloid structure was recently determined by use of X-ray crystallography.<sup>1)</sup> Lounasmaa and Koskinen proposed the biosynthetic route of 1 from un-natural 18-hydroxy-desoxysarpagine (2) as shown in Chart 1.<sup>2)</sup>



R= H : koumine (1)

R= OCH<sub>3</sub> : 11-methoxykoumine (13)

Chart 1

18-Hydroxygardnerine (3) was isolated from leaves of *Gardneria nutans*<sup>3)</sup> ( Log-  
 aniaceae ) and also obtained as a LiAlH<sub>4</sub> reduction product of hydroxygard-  
 nutine.<sup>4)</sup> We report in this communication the biomimetic transformation of 3 to  
 11-methoxykoumine (13) along Lounasmaa's proposal but through a little differ-  
 ent chemical route. We reported already the C/D ring opening reaction of gardnerine  
 (4) with phenyl chlorocarbonate and 4 was converted to 11-methoxytsberpsychine  
 ( = Nb-methylapogardnerine ) (6).<sup>5)</sup> This compound has not been found in nature,

but closely related taberpsychine (7),<sup>6)</sup> a C/D ring opening ether alkaloid, is known.

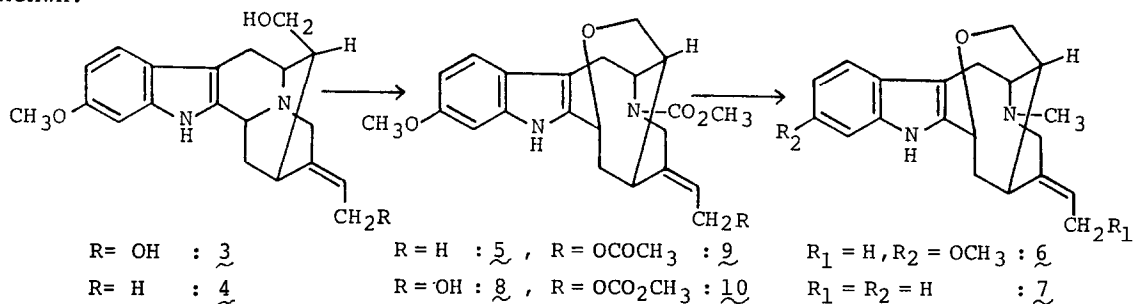


Chart 2

Treatment of 18-hydroxygardnerine (3) with methyl chlorocarbonate ( 2.2 eq mol ) and an excess  $\text{Na}_2\text{CO}_3$  in aq-THF solution gave an amorphous Nb-carbomethoxy-apo-18-hydroxygardnerine (8)<sup>7)</sup> in 80% yield.

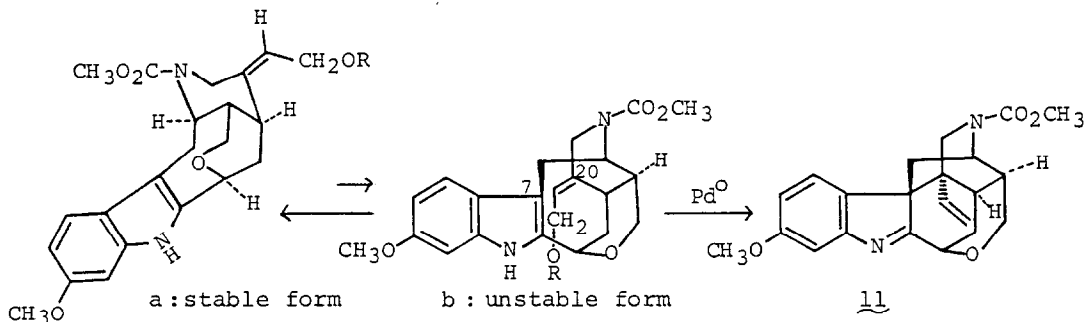
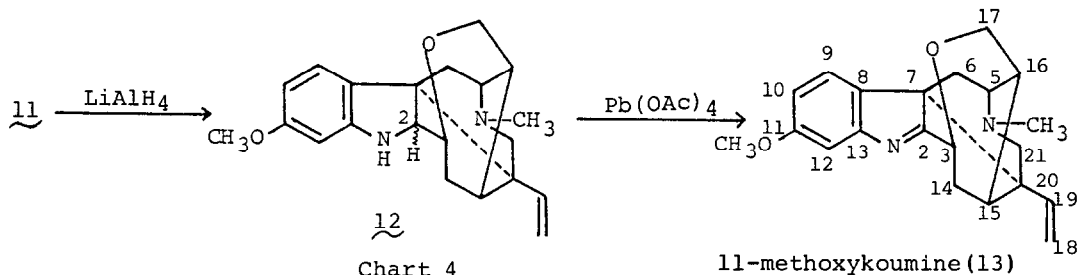


Chart 3

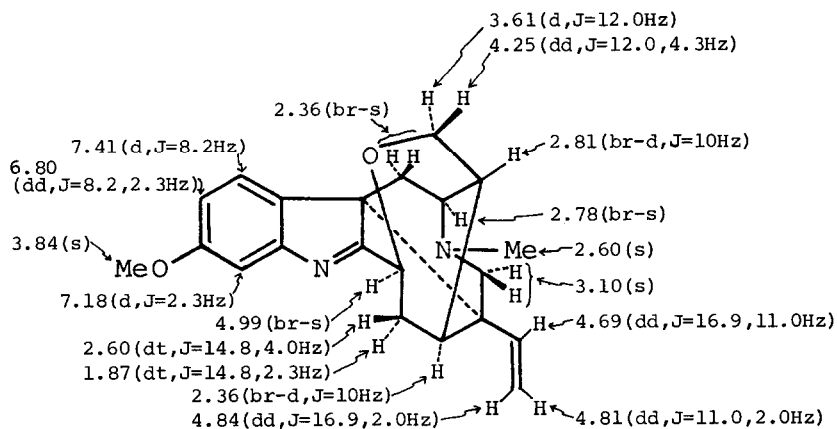
The bond distance between C-7 and C-20 in unstable form 8b shown in Chart 3 is less than 2.5Å from Büchi molecular model examination. Allylation of  $\beta$ -position of indole by use of  $\text{Pd}^0$  catalyst has been reported,<sup>8)</sup> and therefore 8b has possibility of C<sub>7</sub>-C<sub>20</sub> bond formation by above reaction. Allyl alcohol 8 was converted to acetate 9<sup>9)</sup> and methoxy carbonate 10<sup>10)</sup> by usual means. The acetate (9) was brought to indole anion with NaH in DMF solution and in the presence of triphenylphosphine,  $\text{Pd}(\text{OAc})_2$  was added to the DMF solution under Ar. After stirring for 4hr at 80-100°C, reaction mixture was treated with aq- $\text{NH}_4\text{Cl}$  and ether. The residue from organic layer was purified by use of column chromatography. The ring closed compound 11 was obtained in 56% yield in an amorphous state which showed the UV spectrum characteristic to 11-methoxy-indolenine alkaloids<sup>4,11)</sup> and vinyl signals on  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR.<sup>12)</sup> Under the same condition carbonate 10 also gave rise to ring closed compound 11 in the 39% yield. Upon reduction with  $\text{LiAlH}_4$  at rt, 11 was transformed solely into Nb-methylindoline derivative mp 166-168°C (12)<sup>13)</sup> which exhibited the UV spectrum of 11-methoxy-indoline type alkaloids.<sup>4,11)</sup> It was difficult to clarify stereochemistry of C-2 in 12 from NMR spectral analysis but this chiral centre disappeared in the next step. Oxidation of 12 with  $\text{Pb}(\text{OAc})_4$  gave rise to 11-methoxykoumine (13) ( mp 126-127°C, colorless needles, in 78% yield ).<sup>14)</sup>



By the comparison of  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra and CD spectra between partially synthesized 13 and natural koumine 1 which was isolated from *Gelsemium elegantum* in Thailand, the partially synthesized 13 is proved to have the same molecular skeleton and the same absolute configuration with koumine 1.

$^{13}\text{C}$ - and  $^1\text{H}$ -NMR Chemical Shifts and Assignments  
for 11-methoxykoumine (13) (in  $\text{CDCl}_3$ )

No.	(13)
2	186.7 (s)
3	70.9 (d)
5	56.9 (d)
6	28.6 (t)*
7	57.4 (s)
8	135.5 (s)
9	123.2 (d)
10	112.0 (d)
11	160.1 (s)
12	106.7 (d)
13	156.2 (s)
14	25.3 (t)*
15	33.0 (d)**
16	38.8 (d)**
17	61.3 (t)
18	115.8 (t)
19	137.4 (d)
20	45.3 (s)
21	57.8 (t)
N-CH <sub>3</sub>	42.6 (q)
-OCH <sub>3</sub>	55.3 (q)



\* Values may be interchanged.

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- 6) R. H. Burnell and J. D. Medina, *Canad. J. Chem.*, 49, 307 (1971).
- 7) 8: Mass, Calcd. for  $C_{22}H_{26}N_2O_5$ , 398.1839, Found. 398.1831 (base peak).  
IR  $\nu_{\max}^{CHCl_3}$  1670  $cm^{-1}$  (NCOOCH<sub>3</sub>); NMR (270MHz),  $\delta_{ppm}^{CDCl_3}$  5.63 (1H, m, C<sub>19</sub>-H), 5.12 (1H, d, J=0.6Hz, C<sub>3</sub>-H), 3.83 (3H, s, OCH<sub>3</sub>), 3.75, 3.74 (3H, each s, NCOOCH<sub>3</sub>).
- 8) W. E. Billups, R. S. Erbes and L. E. Reed, *Synthetic commun.*, 10, 147 (1980).
- 9) 9: CH<sub>3</sub>COCl/Py. in 77% yield, Mass, Calcd. for  $C_{24}H_{28}N_2O_6$ , 440.1945, Found. 440.1948 (base peak).  $\nu_{CO}^{CHCl_3}$   $cm^{-1}$ , 1730 (OCOCH<sub>3</sub>), 1690 (NCOOCH<sub>3</sub>);  $\delta$ , 3.76, 3.75 (3H, each s, NCOOCH<sub>3</sub>), 2.08 (3H, s, OCOCH<sub>3</sub>).
- 10) 10: CH<sub>3</sub>OCOCl/Py/DMAP; in 44% yield, Mass, 380 (100%), 456 (M<sup>+</sup>, 24%),  $\delta$ , 3.72 (6H, s, OCOOCH<sub>3</sub>, NCOOCH<sub>3</sub>).
- 11) S. Sakai, A. Kubo and J. Haginiwa, *Tetrahedron Let.*, 19, 1485 (1969).
- 12) 11: Mass, Calcd. for  $C_{22}H_{24}N_2O_4$ , 380.1734, Found. 380.1720 (M<sup>+</sup>, base peak),  $\nu_{\max}^{EtOH}$  nm 232, 276, 300 (sh),  $\delta$ , 4.94, 4.91 (dd, J<sub>1</sub>= 17.5Hz, J<sub>2</sub>= 1Hz C=C<sup>Ha</sup>), 4.88 (d, J= 10.6Hz, C=C<sup>Hb</sup>), 4.71 (dd, J<sub>1</sub>= 17.5Hz, J<sub>2</sub>= 10.6Hz, -CH<sub>c</sub>=C ).  
<sup>13</sup>C-NMR,  $\delta$ , 186.0 (2), 70.8 (3), 49.1 (5), 33.8 (6), 56.6 (7), 134.7 (8), 123.1 (9), 112.3 (10), 160.3 (11), 106.9 (12), 156.1 (13), 25.0 (14), 34.0 (15), 39.2, 38.8 (2 signals, 16), 60.6 (17), 116.6, 116.3 (2 signals, 18), 136.4, 136.2 (2 signals, 19), 45.6 (20), 48.7 (21), 55.5 (OCH<sub>3</sub>), 155.3, 52.6 (NCOOCH<sub>3</sub>).
- 13) 12: colorless needles (acetone-hexane), in 67% yield,  $\lambda_{\max}^{EtOH}$  nm (log  $\epsilon$ ), 297 (3.69), 240 (sh, 3.74), 210 (4.56). Calcd. for  $C_{21}H_{26}N_2O_2$ , C, 74.52; H, 7.74; N, 8.28; M<sup>+</sup>, 338.1993, Found. C, 74.31; H, 7.75, N, 8.23; 338.1998 (M<sup>+</sup>, base peak).  $\delta$ , 2.47 (3H, s, N-CH<sub>3</sub>),  $[\alpha]_D^{19}$  -28.0 (c=0.1, MeOH).
- 14) 13:  $\lambda_{\max}^{MeOH}$  nm (log  $\epsilon$ ), 301 (sh, 3.27), 276 (3.50), 232 (4.40). Calcd. for  $C_{21}H_{24}N_2O_2$ , C, 74.97; H, 7.19; N, 8.33; M<sup>+</sup>, 336.1836, Found, C, 74.78; H, 7.26; N, 8.24; 336.1836 (M<sup>+</sup>, base peak);  $[\alpha]_D^{16}$  = -219 (c= 0.19, MeOH); CD,  $\lambda_{\text{peak}}^{MeOH}$  nm ( $\Delta\epsilon$ ), 300 (0), 250 (-15.4), 241 (0), 232 (+13.3), 224 (+5.9), CD of koumine (1), 305 (0), 268 (-11.9), 234 (0), 222 (+21.6), 208 (0), 202 (-3.9).

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